## The Reactivity of Imide Carbonyl Groups in the Intramolecular Aza-Wittig Reaction. An Efficient Route to Iminolactam Derivatives

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Treatment of N-( $\omega$ -azidoalkyl)imides with triphenylphosphine in toluene or xylene under reflux gave the corresponding iminolactam derivatives in good yields via the Staudinger reaction followed by the intramolecular aza-Wittig reaction.

The intramolecular aza-Wittig reaction has attracted considerable attention recently because of its high potential for synthesis of nitrogen heterocycles. 1-3 The ready and clean generation of iminophosphoranes from azides, and their aza-Wittig type reaction with carbonyl groups provides a regiospecific synthesis of imines.<sup>1,2</sup> Several interesting applications of the intramolecular version of the above sequence have appeared recently.3 However, the reactivity criteria for these aza-Wittig reactions seems so far to be ambiguous compared with the Wittig olefination reactions.4 Ester carbonyl groups are unreactive generally in intermolecular aza-Wittig reactions but they react in the intramolecular version to afford the corresponding iminocyclisation products, at least for 5-membered ring formation.5 The intermolecular reaction of phthalic anhydride with iminophosphoranes has been reported recently to afford phthalimides and iminophthalic anhydrides.6 We report here the reactivity of imide carbonyl groups in the intramolecular aza-Wittig reactions, thus providing an efficient route to iminolactam derivatives.

The starting N-( $\omega$ -azidoalkyl)imides (2a—e) were readily obtained from the corresponding bromides (1a-e) by treatment with sodium azide (2.5 fold excess) in a benzene-water two-phase system using a phase-transfer catalyst (Adogen 464 or Aliquat 336) (Table 1).† The reaction of (2) with an equimolar amount of triphenylphosphine (TPP) in toluene occurred spontaneously at room temperature (nitrogen gas evolution by the Staudinger reaction) and ceased during 1 h; however, the cyclisation to give the iminolactams (3) by the aza-Wittig reaction required heating (Scheme 1). For example, (3a) was obtained in 35% yield by heating at 80 °C for 15 h after preparative TLC (silica gel, CHCl<sub>3</sub>-AcOEt). The yield of (3a) increased to 58% on heating at 110 °C for 2 h, and to 93% on heating at 140°C for 4 h. Azides (2b—c) were treated similarly with TPP in toluene, followed by heating to reflux for 2—4 h. Usual work-up and chromatography (silica gel, CHCl3-AcOEt or hexane-AcOEt) afforded the corresponding iminolactams (3b—e) in the yields summarised in Table 1. The iminocyclisation depended on the Y moiety of

Table 1. Yields of azides (2) and iminolactams (3).

Y	<b>(2</b> ), %	(3), %	M.p. of (3), <i>t</i> /°C
a -(CH <sub>2</sub> ) <sub>2</sub> -	65	93	156—157 <sup>b</sup>
<b>b</b> $-(CH_2)_3$ -	98	84	77.5—79
$c - (CH_2)_4$	98	22(59)a	114116.5
<b>d</b> -o-CH <sub>2</sub> C <sub>6</sub> H <sub>4</sub> -	95	99	182—185.5
e -o-CH <sub>2</sub> C <sub>6</sub> H <sub>4</sub> CH <sub>2</sub> -	87	52(89)a	188.5—191

 $<sup>^</sup>a$  In parentheses reaction in xylene under reflux for 4 h.  $^b$  Lit.  $^7$  139—141  $^{\circ}\text{C}.$ 

(2). For example, (2c) having a longer azidoalkyl chain gave only a 22% yield of (3c) under the above conditions, while (2d) and (2e) having o-azidomethyl-phenyl and -benzyl groups respectively gave better yields of the cyclisation products (3d,e) than the corresponding simple alkyl azides. This could be ascribable to the restricted conformation of the side chain in (2d,e), that should be entropically favourable for the cyclisation. The yields of (3c,e) improved to 59 and 89%, respectively by reaction in xylene under reflux for 4 h (Table 1).

The above one-pot iminocyclisation of the *N*-azidoalkyl-phthalimides could be extended to succinimide and glutarimide systems (Scheme 2). The azides (**4a,b**) similarly prepared from the bromides were treated with TPP in xylene at room temperature for 1 h, followed by heating to reflux for 2 h to provide the corresponding iminolactams (**5a,b**), m.p. 151—153 °C and 116—120 °C, respectively, both in 92% yield.

(1) 
$$X = Br$$
(2)  $X = N_3$ 

$$\downarrow \qquad \qquad \qquad \downarrow \qquad \qquad \qquad \downarrow \qquad \qquad \qquad$$

**Scheme 1.** For Y, see Table 1. Reagents and conditions: i, Ph<sub>3</sub>P in toluene, room temperature; ii, heat.

(CH<sub>2</sub>)<sub>n</sub> N (CH<sub>2</sub>)<sub>n</sub> N (CH<sub>2</sub>)<sub>n</sub> N (5)

$$a; n = 2$$

$$b; n = 3$$

**Scheme 2.** Reagents and conditions: i, Ph<sub>3</sub>P in xylene, room temperature, 1 h, heat to reflux, 2 h.

<sup>†</sup> All new compounds described here had spectral and microanalytical properties in agreement with the assigned structures.

Iminolactams are useful intermediates for further modifications of the nitrogen heterocycles. Simple derivatives of (3a) are known as antidepressants.<sup>8</sup>

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